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# EFFECT OF NH<sup>4</sup> ON NITROGENASE ACTIVITY IN NODULE BREIS AND BACTEROIDS FROM *PISUM SATIVUM* L.

#### SEPPO O. SALMINEN

Prairie Regional Laboratory, National Research Council of Canada, Saskatoon, Saskatchewan S7N OW9 (Canada)

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## Summary

Nodule breis and bacteroid preparations were made from Pisum sativum L. (cv. Trapper) inoculated with a single strain of Rhizobium leguminosarum. The detached nodules were triturated under helium flow. The resultant breis could support  $C_2H_2$  reduction in N-Tris[hydroxymethyl]methyl-2-aminoethane-sulfonic acid buffer (Tes) without any additions for over an hour.  $NH_4^+$  was found to inhibit  $C_2H_2$  reduction and  $H_2$  evolution. The inhibition was not dependent on the counterion and was evident immediately after the addition of  $NH_4^+$  to the reaction mixture. L-Methionine-D,L-sulfoximine, added to inhibit assimilation of  $NH_4^+$ , had no effect on the inhibition. Addition of pyruvate enhanced the rate of  $C_2H_2$  reduction in breis and partially overcame the inhibition by  $NH_4^+$ . Pyruvate was found necessary for measurable activity in bacteroid preparations. When ATP and an ATP-generating system were used in breis the effect of  $NH_4^+$  was not observed.

#### Introduction

 $\mathrm{NH_4^+}$  has been shown to repress the synthesis of nitrogenase [1,2] in asymbiotic nitrogen-fixing bacteria and to inhibit the rate of  $\mathrm{C_2H_2}$  reduction [3,4]. A direct inhibition of nitrogen and acetylene reduction in cell-free extracts of Azotobacter vinelandii has also been reported [5], but effects on purified nitrogenase remain to be established in most organisms. Under conditions where free-living Rhizobia have been induced to fix nitrogen, repression by

 $NH_4^+$  has been shown [6]. Tubb [7] interpreted his results as evidence for partial repression or inhibition depending on the method of measurement. Scowcroft et al. [8] found inhibition of nitrogenase activity by  $NH_4^+$ , which could be overcome by  $O_2$ . In symbiotic associations  $NH_4^+$  has also been shown to decrease nitrogen fixation in intact plants [9–11] and in detached nodules [12–15].

However, inhibition of nitrogenase activity by  $NH_4^+$  has not been demonstrated in nodule breis or bacteroid preparations [13–16]. In cell-free *Rhizobium* extracts no effect of  $NH_4^+$  was found [17]. This has led to speculation that in the symbiotic system the inhibition of nitrogen fixation is not by  $NH_4^+$  itself, but by a product of its further assimilation [14]. This study was undertaken to examine the effects of  $NH_4^+$  on nodule breis and bacteroids of *Pisum sativum* L.

### Materials and Methods

Plant Culture. Seeds of P. sativum L. (cv. Trapper) were planted 2 cm deep in 9.5 cm plastic pots of Turface (Wyandotte Chemicals Ltd, Wyandotte, MI). After 3 days the pots were inoculated with 3 ml of rhizobial culture ( $10^9$  cells/ml), which had been incubated for 3 days at  $28^{\circ}$ C in a mannitol/yeast extract medium, pH 7.3 [18]. The bacterial strain used was a mutant selected for resistance to streptomycin ( $50 \mu g/ml$ ) and tetracycline ( $0.25 \mu g/ml$ ). The plants were grown in a controlled environment cabinet (Enconaire System Ltd, Winnipeg, Canada) with a 16 h photoperiod and a day/night temperature of  $20/15^{\circ}$ C. The mean irradiance of  $250 \mu E \cdot m^{-2} \cdot s^{-1}$  was provided by Sylvania cool white fluorescent and incandescent lamps. The watering regime consisted of N-free nutrient solution three times a week and water on other days. The nutrient solution was modified from that used by Wilson and Reisenauer [19] with  $2 \mu g \cdot ml^{-1}$  Fe (Sequestrene 330 Fe, CIBA-GEIGY).

Preparation of nodule breis. Nodules from 31–35-day-old plants were removed and weighed. The nodules (300–400 mg) were triturated in a glove box under helium flow, using a mortar and pestle. The grinding medium (1 ml/70 mg nodules) consisted of 80 mM Tes buffer pH 7.1/1 mM MgCl<sub>2</sub>/200 mM ascorbate/2% (w/v) poly(ethylene glycol) grade 4000/5 mM sodium diethyl-dithiocarbamate/5 mM dithiothreitol. When bovine serum albumin was used its concentration was 1% (w/v). The homogenate was filtered through 2 Miracloth (Chicopee Hills, Inc. Milltown, NJ) discs in a 10 ml syringe and used in brei assays.

Preparation of bacteroids. Bacteroids were isolated by transferring the brei obtained from 500—600 mg nodules into a capped centrifuge tube under helium flow and centrifuging at  $2000 \times g$  for 10 min. The mitochondrial fraction was obtained by centrifuging the supernatant fluid from the bacteroid preparation at  $40\,000 \times g$  for 2 min. The bacteroid and mitochondrial pellets were then resuspended in fresh grinding medium. All transfers and resuspensions were carried out under helium flow.

Acetylene reduction assay. All reactions were carried out in  $1.5 \times 10$  cm test tubes capped with serum stoppers. The final volume of the reaction mixture was 1.5 ml consisting of 0.5 ml brei or bacteroid preparation and 1 ml Tes buffer with other additions as indicated in figure and table legends. All additions

were buffered to pH 7.1. L-Methionine-D,L-sulfoximine, an inhibitor of glutamine synthetase [20], was used at 10 mM final concentration. KCl up to 20 mM had no detectable effect on  $C_2H_2$  reduction but was always included in the controls. The  $O_2$  concentration in the gas phase (10.5 ml) above the reaction mixture was 2% for the breis and 1% for the bacteroids. These were found to be the optimal concentrations in our assay system. The tubes were placed in a Buchler Evapomix shaker at 30°C and shaken for 2 min at 300 rev./min. The assay was then started by injecting  $C_2H_2$  to give a final saturating concentration of 5% (v/v) in the gas phase. Gas samples (0.25 ml) were withdrawn at 10 or 15 min intervals and assayed in a gas chromatograph. At the conclusion of the experiment the pH of the tubes was checked to ensure that it had not changed.

## Results

Addition of ascorbate to Tes buffer grinding medium was found to be necessary for activity. The rate of  $C_2H_2$  reduction was further enhanced by poly(ethylene glycol) which gave higher activity and better yield than solid poly(vinyl pyrrolidone). Poly(ethylene glycol) prevents proteins from binding to tannins [21] and adds to the osmotic potential of the medium stabilizing the bacteroids. Inclusion of 0.3 M sucrose in this grinding medium resulted in lower activity of the breis, and sucrose was, therefore, omitted from the medium. Sodium diethyldithiocarbamate, a copper chelator, added as protection against poly(phenol oxidase), and the sulfhydryl reagent dithiothreitol did not increase activity but were included to promote stability.

The breis could sustain linear rates of  $C_2H_2$  reduction in Tes buffer reaction mixture, without any additions, for over an hour. A lag period was sometimes seen in the beginning of the experiment. Increasing the speed of shaking seemed to minimize the lag period, and sampling at 15, 30 and 45 min ensured that the linear portion was used for calculation of the rates.

The results of Fig. 1 show that NH<sub>4</sub>Cl decreased the rate of  $C_2H_2$  reduction and that the inhibition increased with the concentration of NH<sub>4</sub>Cl. The mean rate (±S.E.) of the control (no NH<sub>4</sub>Cl) samples was  $1.23 \pm 0.30 \,\mu$ mol/g nodule fresh wt. per h. The magnitude of the inhibition varied greatly among preparations, but in all cases increasing the NH<sub>4</sub>Cl concentration also increased the degree of inhibition. This inhibition was significant (P < 0.05) at NH<sub>4</sub>Cl concentrations of 4 mM or higher. The same inhibition was seen in nodule breis from peas inoculated with a commercial inoculum (data not shown) indicating that it was not restricted to the strain used.

This inhibition was not dependent on the counterion (Table I). In an experiment where 10 mM  $NH_4Cl$  inhibited the  $C_2H_2$  reduction by 31%, the same degree of inhibition was obtained with 5 mM  $(NH_4)_2SO_4$ .

The response to  $NH_4^+$  addition was very rapid (Fig. 2). L-Methionine-D,L-sulfoximine was included in the reaction mixtures to inhibit assimilation of  $NH_4^+$ . Good linearity was found between 10 and 30 min, which suggested that the inhibition was immediate following  $NH_4^+$  addition. If the inhibition depended on assimilation of  $NH_4^+$ , a time lag should have existed. Student's t-test indicated that the rates of  $C_2H_2$  reduction, (see legend of Fig. 2) of the samples receiving  $NH_4^+$ , either at the start of the experiment or after 10 min

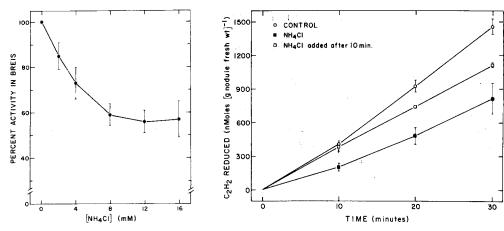


Fig. 1. The relative rates of  $C_2H_2$  reduction by nodule breis as a function of  $NH_4Cl$  concentrations. Each point represents a mean of three to six experiments. The bars represent 2-times S.E. of the mean.

Fig. 2. Acetylene reduced by breis as a function of time. L-Methionine-D,L-sulfoximine was 10 mM. The bars represent 2-times S. E. of duplicate samples. The regression coefficients (rates) for the lines from 10-30 min were  $52.4 \pm 3.0$  ( $\eta$ mol  $C_2H_2/min$ ) ( $\circ$ —— $\circ$ ),  $30.7 \pm 5.8$  ( $\blacksquare$ — $\blacksquare$ ), and  $36.5 \pm 0.9$  ( $\circ$ — $\circ$ ).

were significantly different from the control value (P < 0.01).

Comparison of the effect of  $NH_4^+$  on  $C_2H_2$  reduction and  $H_2$  evolution (Table II) showed that in a brei where  $C_2H_2$  reduction was inhibited by 44% the  $H_2$  evolution also decreased by 51%. Due to the standard error in the control samples in the  $H_2$  measurements the difference between the two percentages was not regarded as significant.

Pyruvate increased the rate of  $C_2H_2$  reduction over the rate of samples in a buffer reaction mixture (Table III). In one experiment  $NH_4^+$  inhibited the rates of buffer samples by 18% and with pyruvate samples the inhibition was only 8%. In another experiment where  $NH_4^+$  concentration was doubled its effect in buffer samples was 66%, whereas the inhibition in samples with pyruvate was

TABLE I

EFFECT OF NH<sub>4</sub>Cl AND (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> ON THE RATE OF ACETYLENE REDUCTION IN BREIS

The reaction mixture consisted of 0.7 ml 71 mM Tes Buffer/0.5 ml brei/0.1 ml 150 mM L-methionine-D,

L-sulfoximine/0.2 ml of the addition given below. The final volume was 1.5 ml of 60 mM Tes, pH 7.1.

Means ± S. E. of duplicate samples.

Addition	Concentration (mM)	C <sub>2</sub> H <sub>2</sub> reduced		
	<b>\</b> ,	$\mu$ mol/g nodule fresh wt. per h	%	
KCl	10	2.29 ± 0.03	100	
NH <sub>4</sub> Cl	10	1.58 ± 0.07	69	
(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	5	$1.57 \pm 0.09$	69	
K <sub>2</sub> SO <sub>4</sub>	5	$2.13 \pm 0.06$	93	

Gas phase	Gas measured	$\mu$ mol/g nodule fresh wt. per h	
,		KCl	NH <sub>4</sub> Cl
He: O <sub>2</sub> , 5% C <sub>2</sub> H <sub>2</sub>	C <sub>2</sub> H <sub>4</sub>	1.69 ± 0.05 (100%)	0.94 ± 0.02 (56%)
He: O <sub>2</sub>	$H_2$	1.52 ± 0.21 (100%)	0.74 ± 0.01 (49%)

only 36%. Thus, pyruvate cut the inhibition by  $NH_4^4$  in half. Bovine serum albumin had no detectable effect on the inhibition. Pyruvate was chosen as the organic acid because it gave consistent results. Although succinate could give higher rates it also often gave lower rates than the buffer controls. This inhibitory effect of succinate has also been reported by Houwaard [15]. We have recently also used malate which gives 3–4-times higher rates of  $C_2H_2$  reduction than pyruvate, without affecting the pattern of  $NH_4^4$  inhibition.

In breis where the inhibition by  $NH_4^+$  was 35 and 14% (Table IV) in buffer reaction mixtures, no inhibition was seen in samples with ATP and an ATP-generating system. In Expt. II, ATP with an ATP-generating system gave lower rates than the buffer reaction mixtures, but still no effect of  $NH_4^+$  was evident. This variable effect of ATP on the rate of  $C_2H_2$  reduction did not seem to depend on the concentration of ATP used.

Only marginal activity could be measured in bacteroid preparations in the absence of exogenous substrates. Addition of pyruvate to the reaction mixture gave measurable linear rates of  $C_2H_2$  reduction. Mitochondria further increased this rate: in an experiment where the rate with pyruvate was  $74.4 \pm 7.6$  nmol/g nodule fresh wt. per h (mean  $\pm S.E.$ ), the rate with pyruvate and mitochondria was  $119.1 \pm 0.5$ . The presence of mitochondria did not appear to change the inhibition by  $NH_4^+$ . Fig. 3 shows that  $NH_4^+$  inhibited  $C_2H_2$ 

Table III  ${\tt EFFECT\ OF\ PYRUVATE\ AND\ NH_4Cl\ ON\ C_2H_2\ REDUCTION\ IN\ BREIS}$ 

The final concentration of pyruvate in the reaction mixture (1.5 ml) was 20 mM. The concentration of NH<sub>4</sub>Cl was 7.5 mM in Expt. I and 15 mM in Expt. II. L-methionine-D,L-sulfoximine was 10 mM. Expt. I had 1% (w/v) bovine serum albumin in the grinding medium. Means  $\pm$  S.E. of duplicate samples.

Experiment	$\mu$ mol C <sub>2</sub> H <sub>4</sub> /g nodule fresh wt. per h				
	Buffer		Pyruvate		
	KCl	NH <sub>4</sub> Cl	KCl	NH <sub>4</sub> Cl	
r	2.50 ± 0.03	2.05 ± 0.03	2.91 ± 0.08	2.69 ± 0.06	
	(100%)	(82%)	(100%)	(92%)	
II	$1.13 \pm 0.08$	$0.38 \pm 0.01$	$1.39 \pm 0.02$	$0.89 \pm 0.12$	
	(100%)	(34%)	(100%)	(64%)	

TABLE IV

EFFECT OF ATP AND ATP-GENERATING SYSTEM ON THE NH<sub>4</sub>Cl INHIBITION OF  $\mathbf{c_{2}h_{2}}$  REDUCTION

The ATP concentrations were 1 mM in Expt. I and 3 mM in Expt. II. Each tube with ATP received 45  $\mu$ mol creatine phosphate and 600  $\mu$ g creatine phosphokinase. The concentration of NH<sub>4</sub>Cl was 7.5 mM. Expt. I was without, Expt. II with 10 mM L-methionine-D<sub>1</sub>L-sulfoximine. Other details as in Table I. Means  $\pm$  S.E. of duplicate samples.

Experiment	$\mu$ mol C <sub>2</sub> H <sub>4</sub> /g nodule fresh wt. per h				
	Buffer		ATP + gen. syst.		
	KCl	NH <sub>4</sub> Cl	KCl	NH <sub>4</sub> Cl	
I	2.09 ± 0.01	1.35 ± 0.06	2.38 ± 0.03	2.86 ± 0.10	
	(100%)	(65%)	(100%)	(120%)	
II	$2.75 \pm 0.18$	$2.36 \pm 0.02$	$1.67 \pm 0.11$	1.77 ± 0.09	
	(100%)	(86%)	(100%)	(106%)	

reduction in bacteroid preparations and that this inhibition increased with NH $_4^4$  concentration. The mean rate of the control samples was 86.8 ± 21.4 nmol/g nodule fresh wt. per h. The replication between different bacteroid preparations was superior to that obtained with nodule breis. The inhibition was significant (P > 0.05) at NH $_4$ Cl concentration of 4 mM or higher.

The activities reported here were rather low. Our more recent data has shown increased rates, especially with malate. Malate required 3% O<sub>2</sub> for optimal activity in our bacteroid system, giving rates in the micromolar range. In an experiment where the rate of C<sub>2</sub>H<sub>2</sub> reduction with pyruvate was  $156 \pm 12$ 

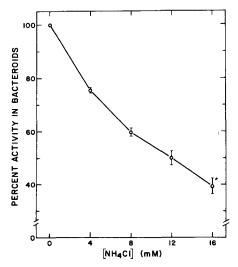


Fig. 3. The relative rates of  $C_2H_2$  reduction by bacteroids as a function of NH<sub>4</sub>Cl concentration. Bovine serum albumin at 1% (w/v) final concentration was used in the grinding medium. Pyruvate was 20 mM, L-methionine-D,L-sulfoximine 10 mM, and 0.1 ml mitochondria (about 50  $\mu$ g mitochondrial protein) were added to the reaction mixture. The points represent the means of six experiments except at 16 mM NH<sub>4</sub>Cl (three experiments). The bars represent 2-times standard errors.

nmol/g nodule fresh wt. per h in the KCl control tubes and  $79 \pm 10$  nmol (51% of the control) with NH<sub>4</sub>, the corresponding rates with malate were  $2.38 \pm 0.13$   $\mu$ mol/g nodule fresh wt. per h and  $1.56 \pm 0.03$   $\mu$ mol (66% of the control) with NH<sub>4</sub>. Other experimental details were as in the legend to Fig. 3. Thus, the inhibition by NH<sub>4</sub> was not a consequence of low activity.

## Discussion

Metabolic regulation of nitrogenase activity in the plant-Rhizobium symbiosis is still poorly understood. The effects of combined nitrogen on intact plants are well documented [9,12,22]. The lapse of a few days between application of nitrogen and observation of the effects make the interpretation of the data difficult in terms of what effects can be contributed to NH<sub>4</sub> itself and which are brought about by its subsequent assimilation. A decrease in leghemoglobin content has been reported following application of 100 mM NH<sub>4</sub>Cl [10]. Evidence for reduction of photosynthesis following 100 mM NH<sub>4</sub>Cl treatment has also been presented [22]. This may be due to uncoupling of photosynthetic electron transport chain [23].

The use of detached nodules was thought to ensure that free  $NH_4^+$  did actually reach the nodules, and an effect on the  $C_2H_2$  reduction was evident within a few hours [14]. The nodules still consist of relatively large amounts of plant host material which could conceivably remove some of the free  $NH_4^+$  or exert other forms of control over the bacteroids in situ. In nodule brei and bacteroid preparations the structural integrity of the host is destroyed allowing a more direct uptake by the bacteroids of the solutes provided. Thus, nodule breis and bacteroid preparations would seem to offer the best opportunity for observing direct, short-term effects of  $NH_4^+$  on the symbiotic partner.

The earlier results reported with nodule breis and bacteroids [15,16] have raised the possibility that free NH<sub>4</sub><sup>+</sup> per se does not exert any control over nitrogen fixation in a symbiotic relationship.

Our data, however, would suggest that at least in P. sativum with our experimental procedures,  $NH_4^*$  does indeed directly inhibit  $C_2H_2$  reduction in breis and bacteroids. This effect was independent of the counterion and was evident within 10 min of the addition of  $NH_4^*$ . As the inhibition by  $NH_4^*$  was also seen in the  $H_2$  evolution it would seem to involve nitrogenase rather than being an artifact of the measuring system. The inhibition was also observed in nodule breis from peas inoculated with a commercial inoculum, suggesting that we are not dealing with a phenomenon restricted to our particular strain of R. leguminosarum.

The data with pyruvate and the ATP-generating system in breis might provide some clues to the perplexing difference between the results of other workers [14,16,24] and our efforts. Pyruvate partially overcame the inhibition by  $NH_4^+$ , and with ATP and the ATP-generating system no inhibition was seen. Customarily these additions are included in the reaction mixtures designed to maximize the rates of  $C_2H_2$  reduction. Our data suggest that this could easily lead to negative results.

In breis, pyruvate cut the inhibition by NH<sub>4</sub> in half. This may also be the case with bacteroid preparations where concentration-dependent inhibition by

 $NH_4^{\dagger}$  was observed. Until we are able to obtain good  $C_2H_2$  reduction in bacteroids without pyruvate, we can not establish the relationship between the effect of  $NH_4^{\dagger}$  and pyruvate in the bacteroid preparation.

The possibility that the effect of  $NH_4^+$  in our system is mediated by its acting as an uncoupler of oxidative phosphorylation remains open. The data with ATP and an ATP-generating system are consistent with this. Pyruvate could act in a phosphoroclastic fashion providing a source of ATP for nitrogenase, and this would be independent of uncoupling. We shall attempt to resolve this question.

The variability we have observed in breis could well be due to the presence of metabolites, enzymes and cofactors from the host material. These undoubtedly contribute to the activity of the breis. Consistent with this was the greater reproducibility among different bacteroid preparations.

In contrast to earlier results reported with detached nodules [14] L-methionine-D,L-sulfoximine did not abolish the inhibition by  $NH_4^+$ . The use of L-methionine-D,L-sulfoximine should slow down assimilation of added  $NH_4^+$  and ensure that the inhibiting agent is indeed  $NH_4^+$  and not an assimilation product. We have detected accumulation of  $NH_4^+$  in the presence of L-methionine-D,L-sulfoximine in breis fixing  $N_2$ .

In this paper, we have demonstrated inhibition of  $C_2H_2$  reduction, as well as  $H_2$  evolution, by  $NH_4^+$  in breis and bacteroids. This inhibition is manifested very quickly after addition of  $NH_4^+$ . It is partially overcome by pyruvate. No effect is seen when ATP and ATP-generating system are included in the reaction mixture. The inhibition by  $NH_4^+$  is consistent with the effects seen in free-living Rhizobia, detached nodules and intact plants. The idea of produced  $NH_4^+$  being able to switch off nitrogenase under conditions leading to high endogenous levels is attractive from the standpoint of high energy cost of nitrogen fixation [22], but the regulatory mechanisms involved remain to be elucidated.

Laane et al. [16] found no effect of  $NH_4^+$  on the energized state of the membrane or on  $C_2H_2$  reduction in *Rhizobium leguminosarum*. The possibility remains that in our system the mode of action of  $NH_4^+$  is analogous to that in *Azotobacter vinelandii* where it switches off the flow of reducing equivalents to nitrogenase by lowering the electrical gradient,  $\psi$ , across the membrane.

At present we do not know how well the added solutes, such as pyruvate, ATP and NH<sub>4</sub>, are actually taken up by the bacteroids. If the membranes are leaky, they would allow for a ready entry of the solutes and would, perhaps, account for the discrepancy between the results of Laane et al. [16] discussed above and ours. From the standpoint of the NH<sub>4</sub> inhibition the integrity of the membranes is not critical, although it will certainly be so in the elucidation of the mechanism of this inhibition.

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